



UNIVERSITY  
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# Ethylene Oligomerisation Catalysts: Synthesis, Evaluation and Mechanistic Studies

by

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Submitted in fulfilment of the requirements for the degree of Doctor of Philosophy

University of Tasmania, April, 2013

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A handwritten signature in black ink, appearing to read 'JA Suttill', written in a cursive style.

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April, 2013

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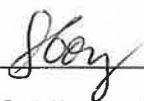
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## Acknowledgements

First, I would like to thank my supervisor Dr. David McGuinness for his support throughout this PhD. Dave has been nothing but patient, encouraging and knowledgeable over the years. Next, I would like to thank Dr. Michael Gardiner for his input into this work and the hours he has spent collecting and refining the crystal data presented herein. I am especially grateful to Prof. Dr. Peter Wasserscheid at Lehrstuhl für Chemische Reaktionstechnik for his supervision and providing me with the opportunity to visit Germany during 2011. I would also like to thank Jens Burkhardt and Nike Agel for looking after me during this time. I wish to thank all the members of 306 for the many conversations and laughs shared during my time here. I would like to acknowledge several members of the CSL for their technical assistance throughout this project. Noel Davies for his invaluable help with product identification and the hours he spent optimising the GC separations, Thomas Rodemann for his work with elemental analysis and the discussions on the best method for purification and finally James Horne for his NMR work.

I am grateful to the University of Tasmania, School of Chemistry at UTas, Sasol Technology, Australian Research Council, the Australian Synchrotron and the Lehrstuhl für Chemische Reaktionstechnik at the Friedrich-Alexander Universität Erlangen-Nürnberg for their assistance.

Finally, I would like to thank my friends and family for their support over the last three years. Without their encouragement this would never have been possible.

## Abstract

The oligomerisation of ethylene to linear alpha olefins is an important industrial process for the production of polyethylene co-monomers, plasticizers, surfactants and waxes. With the ever increasing market demand for short chain linear alpha olefins (1-butene, 1-hexene and 1-octene) there has been intensive research in both industry and academia on methods to selectively oligomerise ethylene to form only the short chain fractions. As such, a range of novel catalysts based on titanium and chromium have been synthesised and characterised. Their behaviour in ethylene oligomerisation and polymerisation has been studied. In most cases, the mechanism of oligomer formation has been investigated.

A range of aryloxy- and alkoxy- ligands, both monodentate and chelating, have been reacted with Ti(IV) to give complexes of the form  $[\text{Ti}(\text{OAr})_2\text{Cl}_2]$ ,  $[\text{Ti}(\text{O}^{\wedge}\text{OR})\text{Cl}_3]$  and  $[\text{Ti}(\text{O}^{\wedge}\text{OR})_2\text{Cl}_2]$  (R = aryl, alkyl). X-ray crystallography of selected samples has revealed that the complexes vary in the Lewis base solvation and/or aggregation state. The complexes have been evaluated as ethylene oligomerisation catalysts in conjunction with alkylaluminium activators such as triethylaluminium and methylaluminoxane. Polymerisation is the major process, however, ethylene oligomers such as dimers and higher oligomers also result. The bimodal distribution of products indicates a number of active species arise when the catalyst is activated. Deuterium labelling studies are consistent with a metallacycle mechanism for oligomer formation.

The mechanism for the catalytic dimerisation of ethylene with the industrially relevant Alphabutol-type catalysts has been explored. Catalysis with  $\text{Ti}(\text{OBu})_4/\text{AlMe}_3$  and a mixture of ethylene/perdeuteroethylene, followed by isotopomer analysis via GC-MS, has revealed the major product 1-butene and the ethylene/1-butene codimers result from a Cossee mechanism rather than the commonly reported metallocycle mechanism.

A range of unsubstituted and 6-substituted *ortho*-phenoxy-imine ligands have been prepared and converted to their silyl ether derivatives. Reaction of silyl ethers with  $\text{TiCl}_4(\text{thf})_2$  in the case of the unsubstituted species yields bis-ligated complexes while the substituted species react cleanly to yield complexes of the form  $[\text{Ti}(\text{O}^{\wedge}\text{NR})\text{Cl}_3(\text{thf})]$ . In most cases the complexes have been characterised by X-ray crystallography. Testing of the complexes for ethylene oligomerisation and polymerisation has been undertaken employing alkylaluminium co-catalysts. In all cases the predominant product formed is polyethylene. However, careful analysis of the liquid phase reveals a complex process by which 1-butene is most likely formed via a Cossee mechanism while 1-hexene results from a metallocyclic process.

A series of mono-anionic tridentate phenoxyimine (3-(*t*-butyl)-2-(OH)- $\text{C}_6\text{H}_4\text{C}=\text{N}(\text{C}(\text{CH}_3)_2\text{CH}_2\text{OMe})$ , 3-(adamantyl)-2-(OH)- $\text{C}_6\text{H}_4\text{C}=\text{N}(2'-(2''\text{-SMe})\text{C}_6\text{H}_4)\text{-C}_6\text{H}_4$  and 3-(*t*-butyl)-2-(OSiMe<sub>3</sub>)- $\text{C}_6\text{H}_4\text{C}=\text{N}(\text{C}(\text{CH}_3)_2\text{CH}_2\text{OMe})$  and phenoxy-amine (3-(adamantyl)-2-(OH)- $\text{C}_6\text{H}_4\text{CH}_2\text{-N}(2'-(2''\text{-OMe})\text{C}_6\text{H}_4)\text{-C}_6\text{H}_4$ ) ligands have been prepared and reacted with  $\text{TiCl}_4$  or  $\text{TiCl}_4(\text{thf})_2$  to give the mono-ligated complexes. In most cases, the solid state structure of the complexes has been determined. The complexes have been tested for ethylene oligomerisation/polymerisation in

conjunction with methylaluminoxane. They show a broad range of selectivities, which are discussed in relation to their ligand structure.

Finally, three classes of ligands have been explored for ethylene tri-/tetramerisation in conjunction with chromium and both triethylaluminium and methylaluminoxane co-catalysts. Hydrazine based ligands containing an N-H functionality [PN(NH)P], analogous to Rosenthal's previously reported PNPNH system, show selectivity towards 1-hexene and 1-octene formation in conjunction with triethylaluminium, and act as PNP tetramerisation analogues when methylaluminoxane is employed. PNP ligands containing non-protic pendant donor moieties generally show poor activity and selectivity when triethylaluminium is employed as an activator, however when methylaluminoxane is used good activities and selectivities are achieved. When oxygen is the donor atom *n*-propylcyclopentane and 2-propenylcyclopentane, and higher homologues, are produced during catalysis. The formation of such products is discussed with respect to the generation of methylenecyclopentane and methylcyclopentane by PNP based tetramerisation catalysts. Simple phosphine ligands containing O-H functionalisation are also explored and it was shown that the catalyst selectivity is highly dependent on both the activator and structural features of the ligand employed.



## Table of Contents

Declaration	I
Acknowledgements	II
Statement of Co-Authorship	III
Abstract	V
Table of Contents	VIII
Abbreviations	XII
<b>Chapter 1</b> Introduction	1
1.1 Ethylene oligomerisation	1
1.2 The discovery of selective trimerisation	6
1.3 Titanium based trimerisation systems	10
1.3.1 Cyclopentadienyl-arene titanium catalysts	10
1.3.2 Tridentate phenoxy-imine titanium catalysts	13
1.4 PNP based tri-/tetramerisation systems	14
1.5 Mechanistic aspects	16
1.5.1 Oxidation state and structural studies	17
1.5.2 Deuterium labeling studies	21
1.5.3 Computational studies	23
1.6 Aims	24
1.7 References	27
<b>Chapter 2</b> Preparation and structures of aryloxy- and alkoxy-Ti(IV) complexes and their evaluation in ethylene oligomerisation and polymerisation	35
2.1 Introduction	35
2.2 Complex synthesis	37
2.3 Structural analysis	38

2.4 Ethylene oligomerisation and polymerisation employing alkoxy complexes	45
2.5 Ethylene oligomerisation and polymerisation employing aryloxy complexes	48
2.6 Mechanistic study	50
2.7 Conclusions	52
2.8 Experimental	53
2.9 References	64
<b>Chapter 3</b> The mechanism of ethylene dimerisation by $\text{Ti}(\text{OR}')_4/\text{AlR}_3$	68
3.1 Introduction	68
3.2 Analysis of 1-butene	73
3.3 Analysis of $\text{C}_6$ oligomers	83
3.4 Conclusions	95
3.5 Experimental	96
3.6 References	100
<b>Chapter 4</b> Ethylene polymerisation and oligomerisation with arene-substituted phenoxy-imine complexes of titanium: Investigation of multi-mechanism behaviour	106
4.1 Introduction	106
4.2 Synthesis and structures	108
4.2.1 Ligand synthesis	109
4.2.2 Complex synthesis and structural analysis employing ligands <b>6</b> and <b>7</b>	109
4.2.3 Complex synthesis and structural analysis employing <i>tert</i> -butyl substituted ligands	114
4.3 Ethylene oligomerisation and polymerisation	118

4.4 Mechanistic study	121
4.5 Investigation of <b>15</b> with triethylaluminium cocatalyst	124
4.6 Conclusions	127
4.7 Experimental	128
4.8 References	143
<b>Chapter 5</b> Synthesis of Ti(IV) complexes of donor-functionalised phenoxy-imine tridentates and their evaluation in ethylene oligomerisation and polymerisation	149
5.1 Introduction	149
5.2 Synthesis and structures	151
5.3 Attempted preparation of ether bridged complexes	159
5.4 Ethylene oligomerisation and polymerisation	162
5.5 Conclusions	169
5.6 Experimental	170
5.7 References	185
<b>Chapter 6</b> Synthesis of a series of simple phosphine and backbone functionalised PNP ligands for evaluation as selective oligomerisation catalysts in conjunction with AlEt <sub>3</sub> and MAO	190
6.1 Introduction	190
6.2 Ligand synthesis	193
6.3 Attempted preparation of hydroxyl functionalised PNP ligands	197
6.4 Ethylene oligomerisation and polymerisation	205
6.4.1 Ligands containing N-H functionality	206
6.4.2 PNP ligands containing pendant donor functionalities	212
6.4.3 Phosphine ligands containing hydroxyl functionality	213
6.4.4 Screening of ligands in conjunction with MAO	215

6.5 Conclusions	219
6.6 Experimental	220
6.7 References	234
<b>Chapter 7</b> Conclusions	240
7.1 General summary	240
7.2 Alkoxy- and aryloxy- titanium complexes	240
7.3 Mechanistic investigations of $\text{Ti}(\text{OR}')_4/\text{AlR}_3$	241
7.4 Arene-substituted phenoxy-imine complexes	241
7.5 Tridentate phenoxy-imine complexes	242
7.6 Phosphine and PNP derived ligands	243
Appendices	244
CIFs	CD

## Abbreviations

Ar	Aryl
CDCl <sub>3</sub>	Deuterated chloroform
DCM	Dichloromethane
Et	Ethyl
GC	Gas chromatography
GC-MS	Gas chromatography-Mass spectrometry
GTL	Gas-to-liquid
HDPE	High density polyethylene
LAO	Linear alpha olefin
LDPE	Low density polyethylene
LLDPE	Linear low density polyethylene
MAO	Methylaluminoxane
Me	Methyl
MS	Mass spectrometry
NMR	Nuclear magnetic resonance
PE	Polyethylene
THF	Tetrahydrofuran
TBAF	Tertbutylammonium fluoride
TON	Turn-over number